#### Notes:

Exam: - Friday Oct. 29, 6 PM, Room 205 MPHY.

<u>Coverage</u> through section 6.4. You can make one page formula sheet. 8.5\*11 inch, both sides.

**Canvas grades & lecture links:** sample exams on Canvas now have solutions.

**Tomorrow** I intend to review; I will accept requests for previous questions from HW etc. to address.

## Recall, Canonical Ensemble (ch. 16):

 $T_1 = constant$ 



<u>At equilibrium</u>, 2 sides can exchange small amounts of energy. Consider 2 specific microstates in system 2:

$$Q = -\Delta U$$

 $\frac{P_A}{P_B} = e^{[S_R(A) - S_R(B)]/k_B} = e^{-\Delta E/k_B}$ Ratio of
<u>multiplicities</u>

$$P_i = \frac{1}{Z} e^{-E_i/kT}$$

Probability of finding system in state *i* in equilibrium (one microstate)

$$Z = \sum_{\text{states } i} Exp[-E_i/kT]$$

**Partition function** 

### Example: spin-1/2 non-interacting paramagnet (revisit in canonical ensemble):

- $E = \pm \mu B$  per atom ->  $E = \mu B(N^{-} N^{+}) = \mu B(2N^{-} N)$  $Z_i = e^{+\mu B}/_{kT} + e^{-\mu B}/_{kT}$  $\frac{N^-}{N^+} = e^{-2\mu B}/_{kT}$  $= 2 \cosh^{\mu B} / _{kT} = 2 \cosh \beta \mu B$ • We derived this in thermodynamic limit (I displayed same formula temperature T. before; microcanonical case). • For constant-*T* situation
- this is replaced by probabilities.

This is Z for system = oneatom contacting the rest of the paramagnetic spins at

### Also showed:

• Energy averaging

$$E_i \rightarrow \langle E \rangle = "U" = -\frac{\partial}{\partial \beta} lnZ = kT^2 \frac{\partial}{\partial T} lnZ$$

 $\beta = \frac{1}{kT}$ 

- average E equivalent to internal energy U in thermodynamic limit
- can show, energy *fluctuations* vanish in thermodynamic limit.
  - Works for small *or* large system; contact with infinitely large reservoir maintains the temperature.
  - Results for canonical ensemble approach those for microcanonical in large-N limit.
  - ▷ However internal energy itself is not a conserved quantity.

(revisit in canonical ensemble):



$$Z_i = e^{+\mu B/_{kT}} + e^{-\mu B/_{kT}}$$
$$= 2\cosh^{\mu B}/_{kT} = 2\cosh\beta\mu B$$

 $Z = \prod_i Z_i$ ; <u>includes all cross terms</u>.

system of distinguishable particles, non-interacting.

$$\frac{P^{-}}{P^{+}} = e^{-\frac{2\mu B}{kT}} = e^{-\frac{2\mu B}{kT}} = 2\cosh\frac{\mu B}{kT} = 2\cosh\frac{\beta \mu B}{kT}$$

 $Z = \prod_i Z_i$ ; <u>includes all cross terms</u>.

solving:

 $\langle E \rangle = -\mu BNtanh(\mu B/kT)$  $C = \frac{\partial \langle E \rangle}{\partial T} = Nk_B \frac{(2\mu B/kT)^2}{(e^{\mu B/kT} + e^{-\mu B/kT})^2}$ 

<u>average</u> values for the entire system (not fixed as in microcanonical ensemble).



### **Canonical ensemble defined quantities:**

$$F \equiv -k_B T ln Z$$

- Free energy can treat as defined quantity.
- F is a conserved quantity, doesn't fluctuate.
- Equivalent to thermodynamic F, we will see.

Then other quantities follow as before:  $(dF = -SdT - PdV + \mu N)$ 

$$S = -\left(\frac{\partial F}{\partial T}\right)_{V,N}$$
  $P = -\left(\frac{\partial F}{\partial V}\right)_{T,N}$   $\mu = \left(\frac{\partial F}{\partial N}\right)_{T,V}$ 

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$$M = k_B lnZ + k_B T \frac{\partial}{\partial T} lnZ$$

$$P = k_B T \frac{\partial}{\partial V} lnZ$$

$$Z = Z(T, V, N)$$

$$\mu = \cdots$$

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$$S = k_B lnZ + k_B T \frac{\partial}{\partial T} lnZ$$

$$\implies F + TS = k_B T^2 \frac{\partial}{\partial T} lnZ = \langle E \rangle$$

solving:

 $\langle E \rangle = -\mu BNtanh(\mu B/kT)$ 

 $F = -NkTln(2\cosh\beta\mu B)$ 

$$S = -\left(\frac{\partial F}{\partial T}\right)_{V,N} \qquad \mu = \left(\frac{\partial F}{\partial N}\right)_{T,V}$$

#### Alternate derivation:



Before, 2 <u>specific microstates</u> in system 2. (microstates have S = 0)  $\frac{P_A}{P_B} = e^{[S_R(A) - S_R(B)]/k_B} = e^{-\Delta E/k_B}$ 

<u>Alternative</u>: system 2 in <u>macrostate</u> *j* with energy  $E_j$ (or perhaps  $E_j \pm \Delta E$ ). Probability of *j*:

$$f_{j} = \frac{\Omega_{res}(subsystem \ has \ E_{j})}{\Omega(all \ possible)} = \frac{exp(\{TS_{res}(U_{sys}) - (E_{j} - U_{sys})\}/kT)}{exp(\{TS_{res}(U_{sys}) + TS(U_{sys})\}/kT)}$$

True in thermo. limit

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#### More on entropy:

$$F \equiv -k_B T ln Z$$

- Free energy can treat as defined quantity.
- *F* is a conserved quantity, doesn't fluctuate.
- Equivalent to thermodynamic F, we will see.

Then: 
$$S = -\left(\frac{\partial F}{\partial T}\right)_{V,N}$$
  $(dF = -SdT - PdV + \mu N)$ 

$$\implies S = k_B lnZ + k_B T \frac{\partial}{\partial T} lnZ$$

From here can show:

$$S = -k_B \sum_{j} P_j \ln(P_j)$$

<u>Gibbs</u> version of entropy, vs. <u>Boltzmann</u> version appropriate for Microcanonical:

$$S = k_B \ln(\Omega)$$